

Short-range structural order in
Zr-based multi - component
glasses, using XAFS

DEBDUTTA LAHIRI

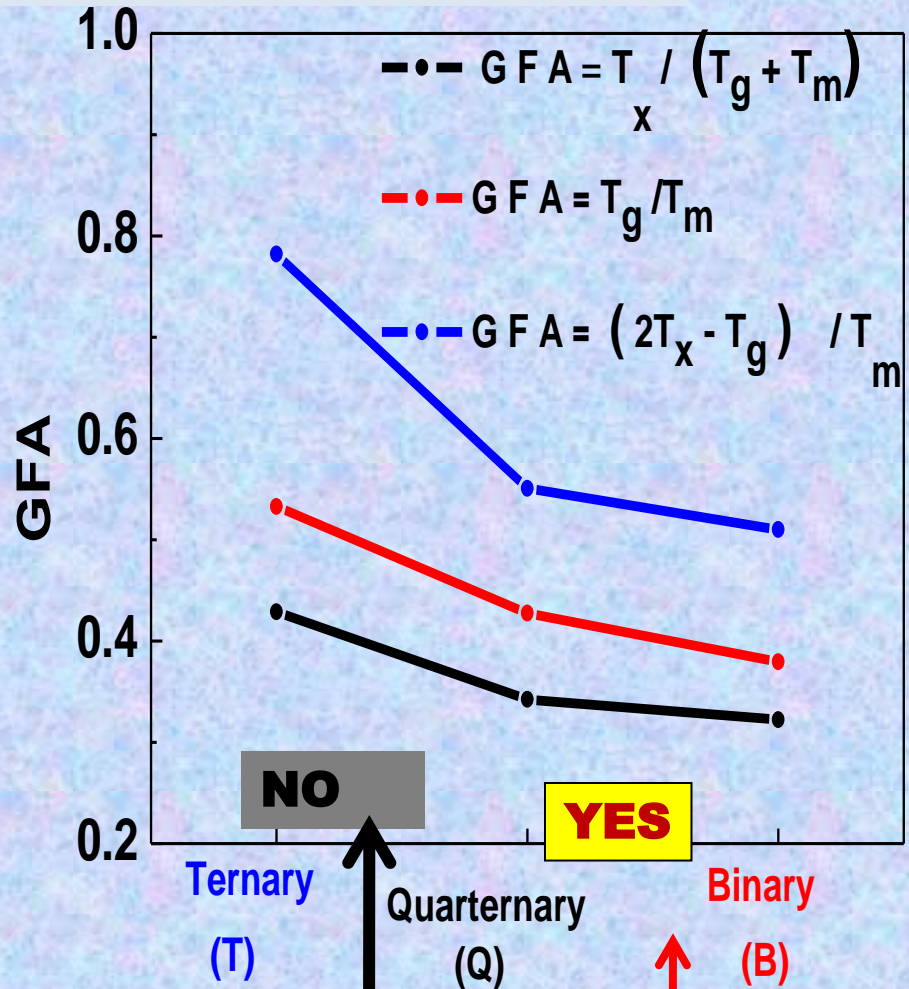
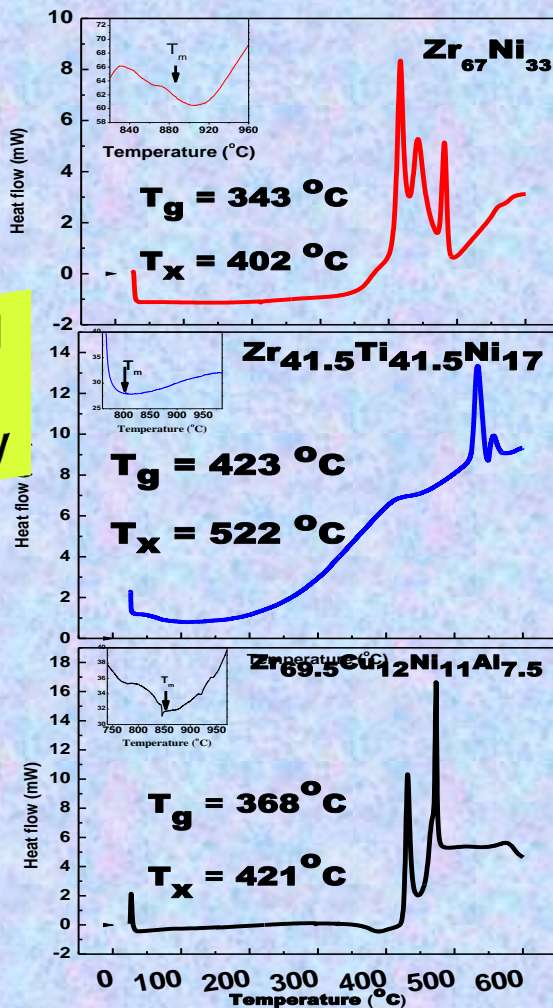
**HIGH PRESSURE & SYNCHROTRON RADIATION PHYSICS
DIVISION**

**BHABHA ATOMIC RESEARCH CENTRE
(BARC), MUMBAI**

GLASS FORMING ABILITY

Zr_{41.5}Ti_{41.5}Ni₁₇ (T) > **Zr_{69.5}Cu₁₂Ni₁₁Al_{7.5} (Q)** > **Zr₆₇Ni₃₃ (B)**


Differential scanning calorimetry



CONFUSION PRINCIPLE

Synthesis of metallic glass ribbons

Alloys: **Zr_{69.5}Cu₁₂Ni₁₁Al_{7.5}**
Zr_{41.5}Ti_{41.5}Ni₁₇
Zr₆₇Ni₃₃

Vacuum arc melting  Melt spinning

Parameters

Wheel Speed – 55 m/s

Wheel to crucible gap ~ 1 mm

Ar gas pressure – 1 kg/mm²

Melt spinning



15-30 mm



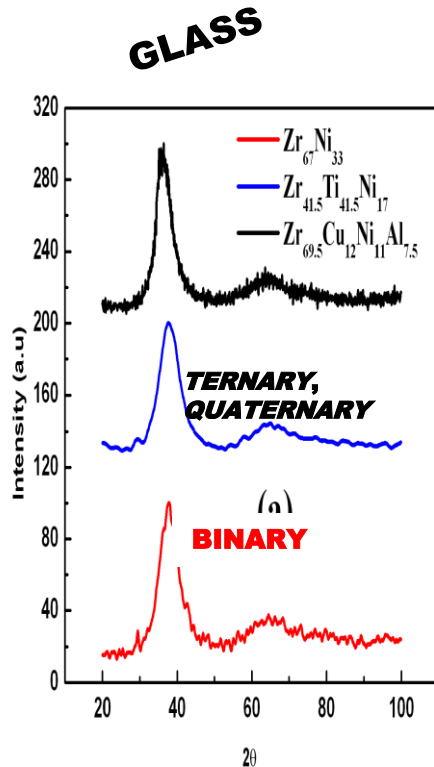
~20 μ

Rapidly solidified metallic glass

GLASS FORMING ABILITY STRUCTURE



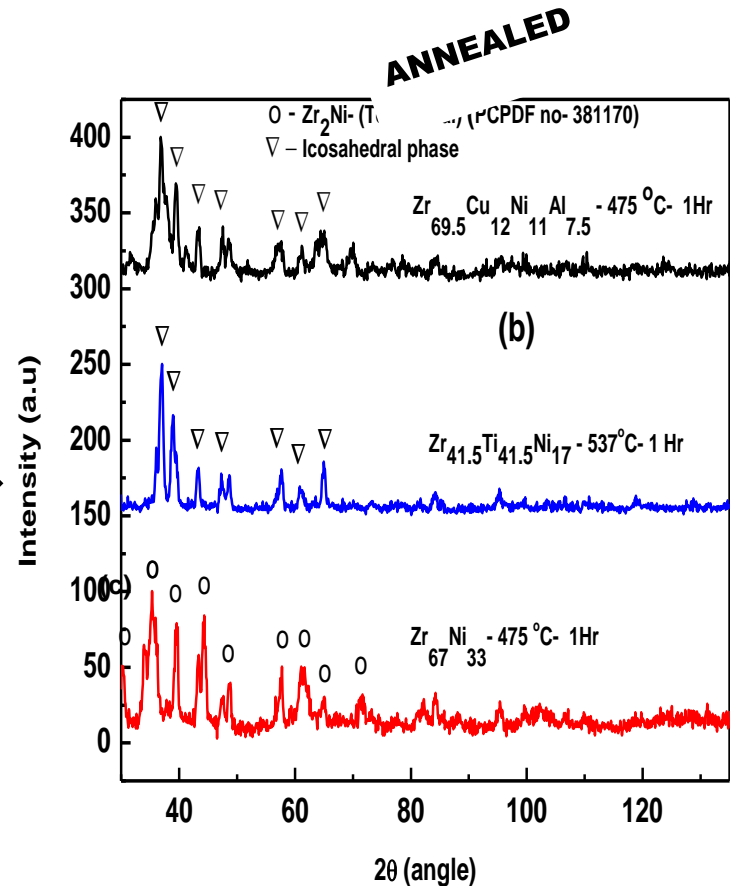
XRD



**QUASI
- CRYSTAL**



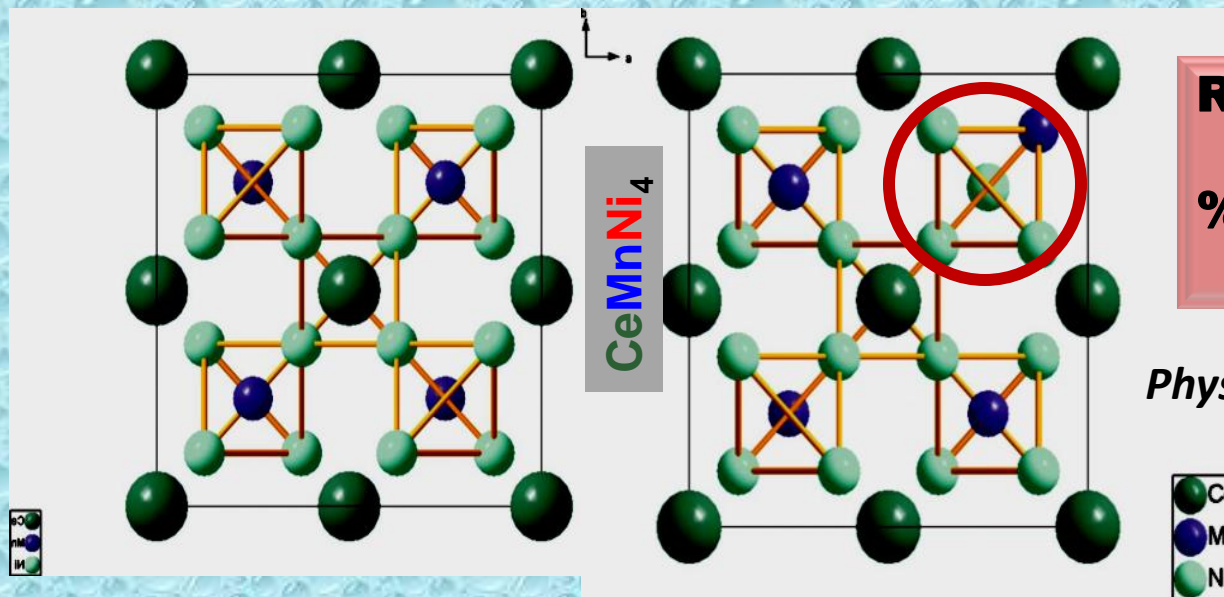
NiZr₂



NO explanation of GFA from LRO

SHORT - RANGE - ORDER (SRO)

**Local structure \neq
Long-Range order (LRO)**



$R_{\text{DEFECT}} < 8 \text{ \AA}$ (SRO)

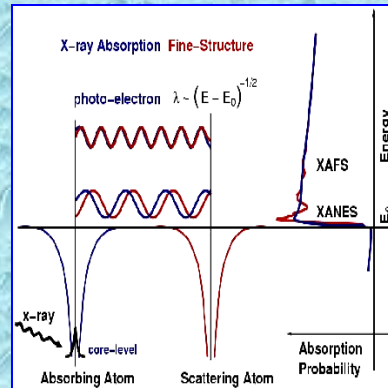
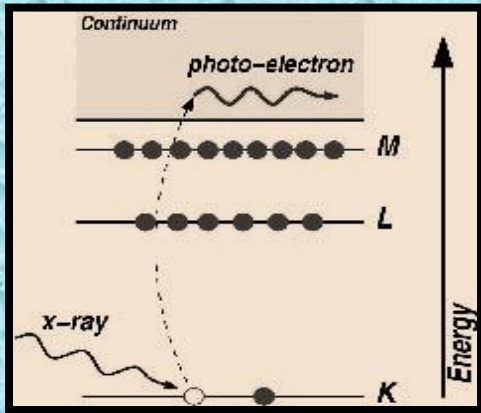
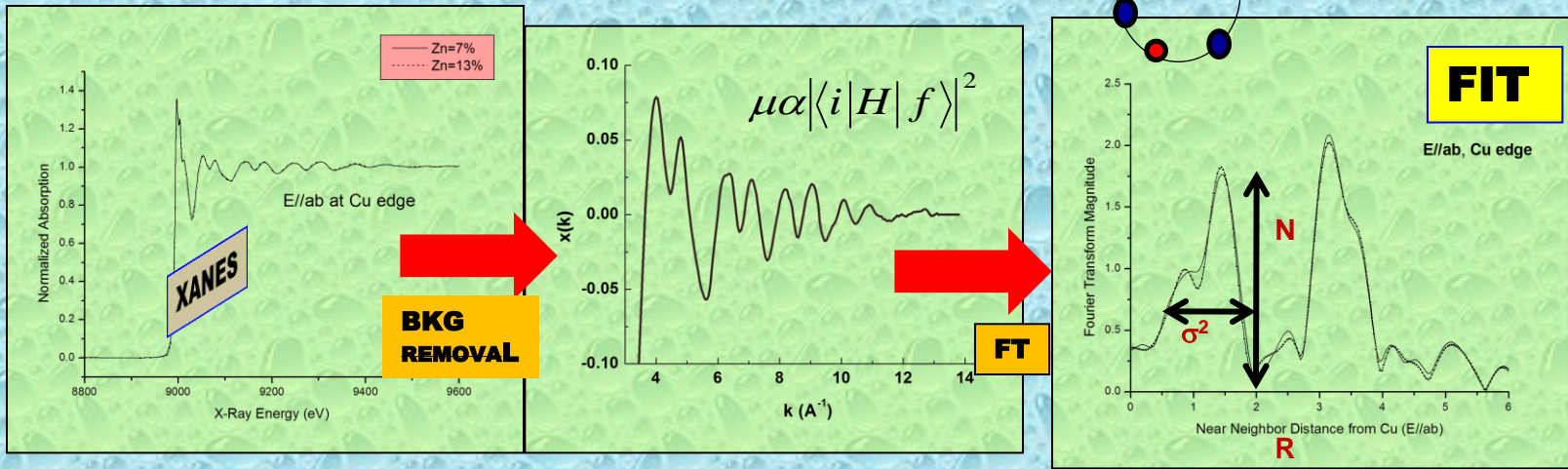
$\%_{\text{DEFECT}} \geq 1\text{-}5\%$

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XAFS ...

XAFS (X-Ray Absorption Fine Structure)



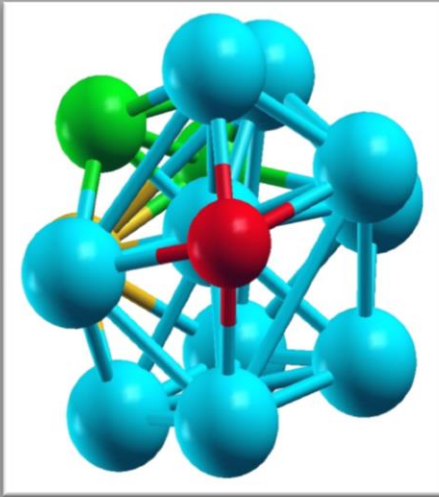
MODEL NOT MUST
FIT AS-IT-IS (error <5%)

COMPLICATED SYSTEMS

-COMPETITIVE MODELS
- SUPPORTING THEORY /
TECHNIQUES

SYNCHROTRON - BESSY (GERMANY), APS (USA)

SRO= CLUSTER



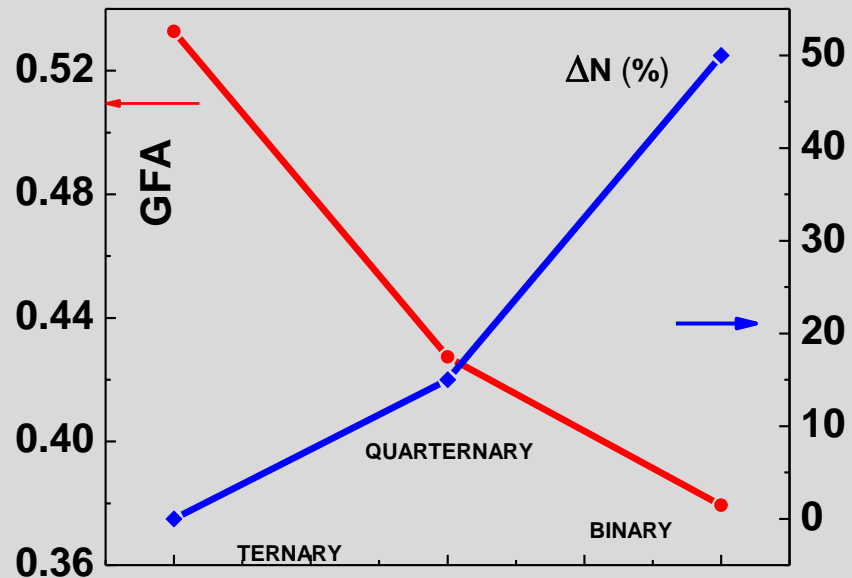
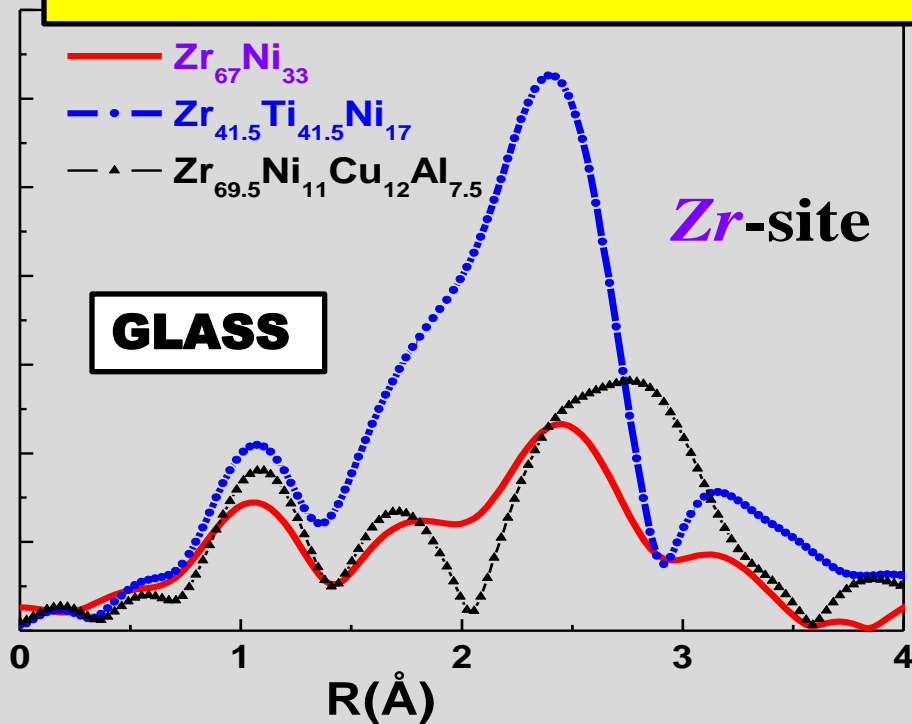
- ❑ **CONFIGURATION**
 - **ICOSAHEDRAL** (ISRO : f_{ISRO})
- ❑ **HETEROGENEITY** (Δf)
- ❑ **DEGREE OF DISORDER** σ^2
- ❑ **CHEMICAL ORDER**
 - **CHARGE TRANSFER**
 - **BOND-LENGTH**

XAFS

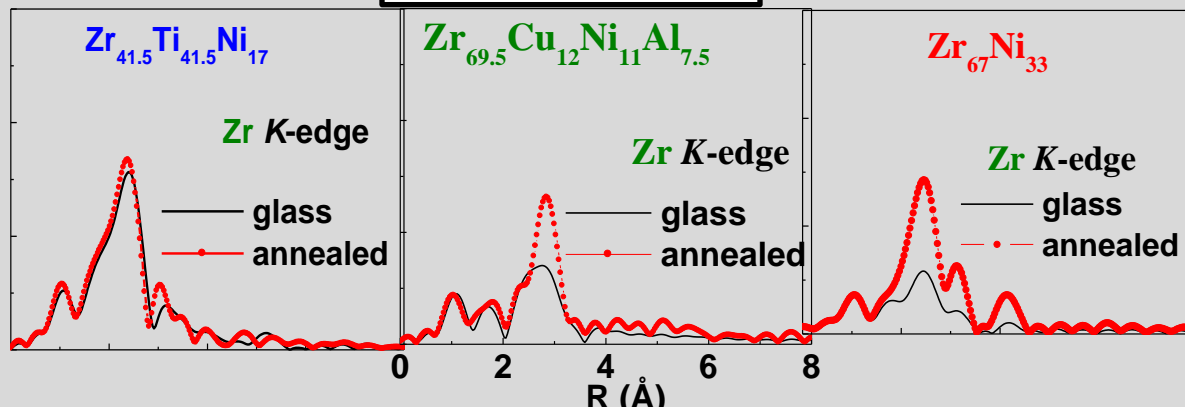
MOLECULAR DYNAMICS + DOS

GLASSY + ANNEALED PHASES

XAFS RESULTS



ANNEALING



MULTI-COMPONENT

**DISORDER
TOO MANY ELEMENTS**

BOND - LENGTHS

XAFS

$$R_{\text{Ni-Zr}} = 2.61 \text{ \AA}$$

$$R_{\text{Ni-Ni}} = 2.61 \text{ \AA}$$

$$R_{\text{Cu-Zr}} = 2.72 \text{ \AA}$$

$$R_{\text{Zr-Zr}} = 3.07 \text{ \AA}$$

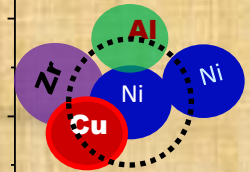
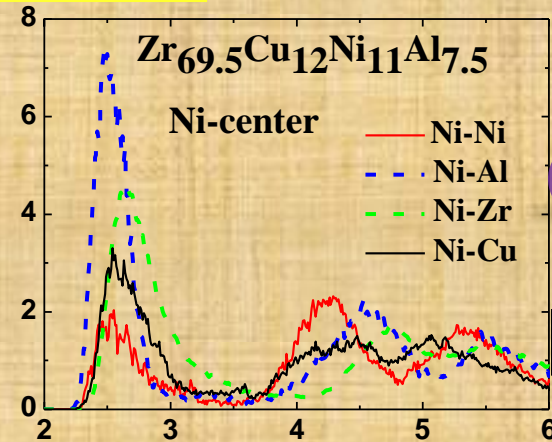
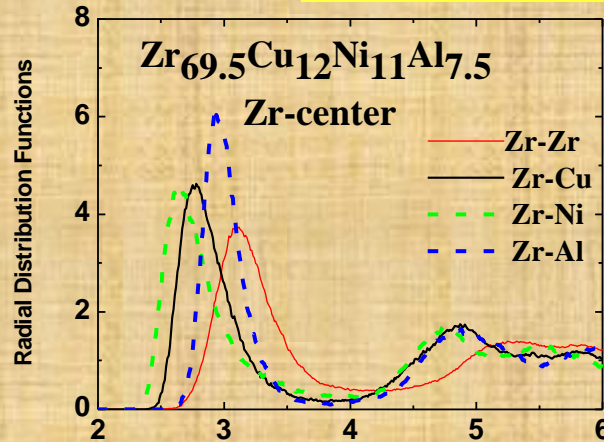
$$R_{\text{Ni-Zr}} = 3.11 \text{ \AA}$$

$$R_{\text{Ni-Ti}} = 2.56 \text{ \AA}$$

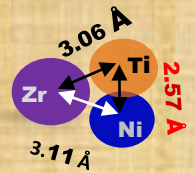
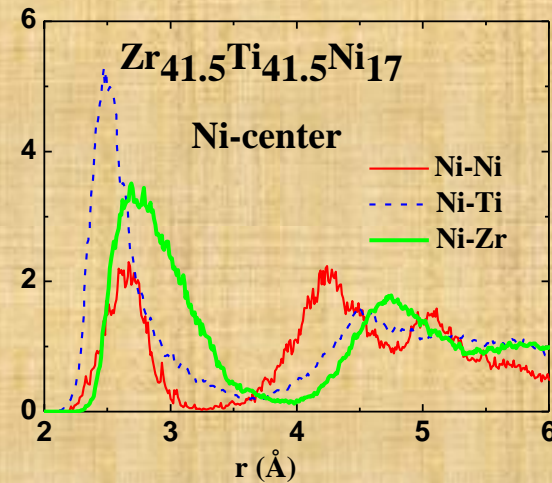
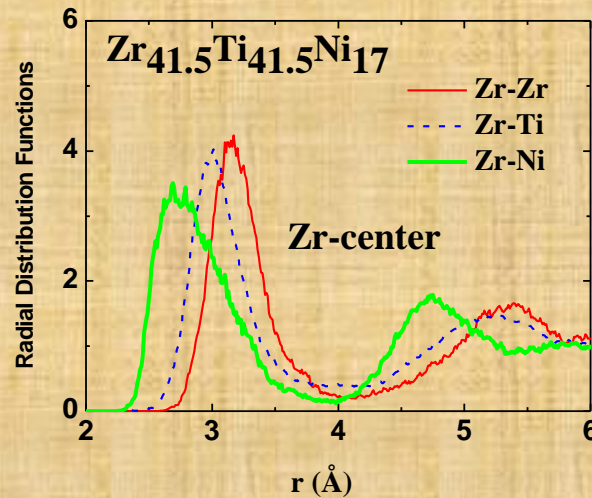
$$R_{\text{Ti-Zr}} = 3.06 \text{ \AA}$$

$$R_{\text{Zr-Zr}} = 3.24 \text{ \AA}$$

MOLECULAR DYNAMICS



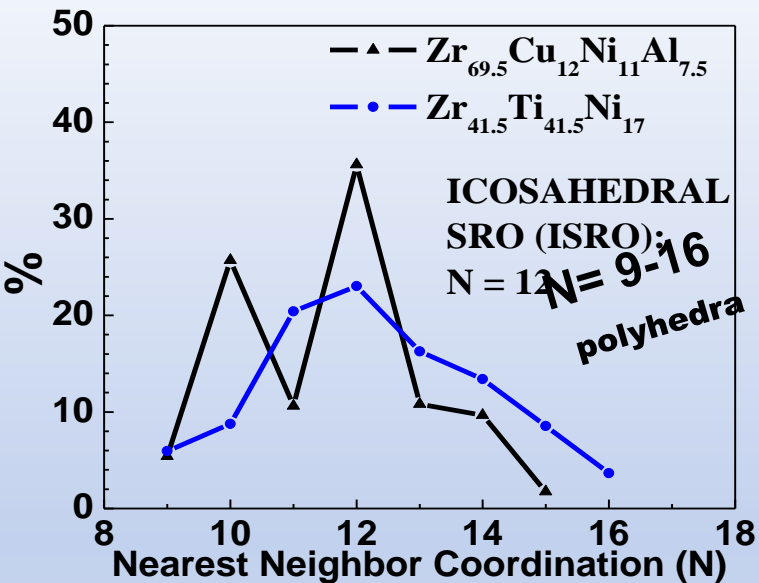
SAME bond-length



$\Delta R = 0.5 \text{ \AA}$

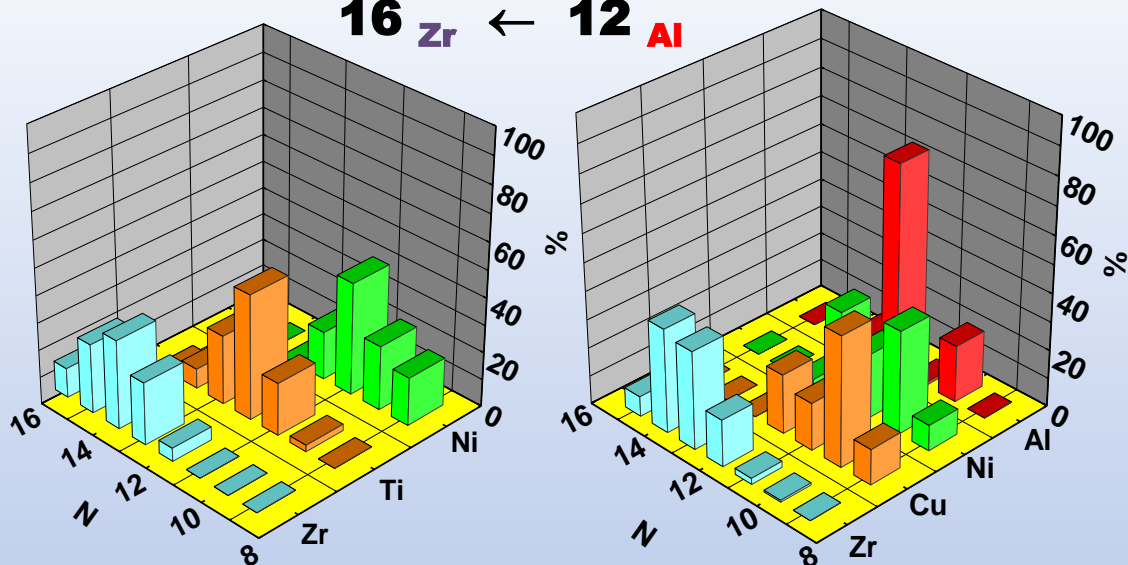
NEAREST NEIGHBOR CONFIGURATION

AIMD SIMULATIONS

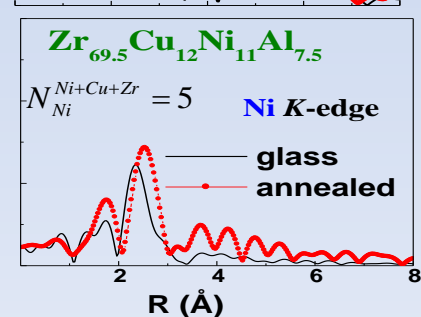
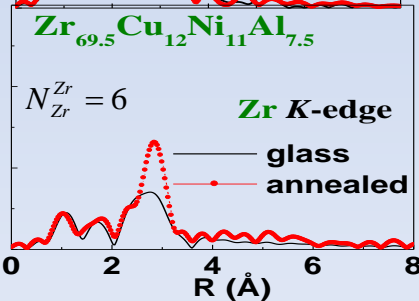
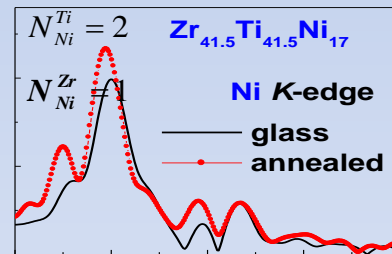
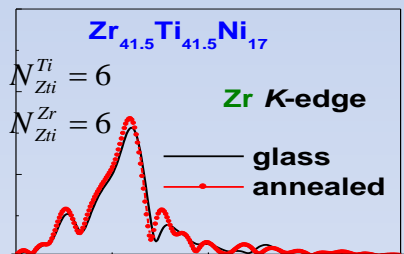


HETEROGENEOUS

16 Zr ← 12 Al



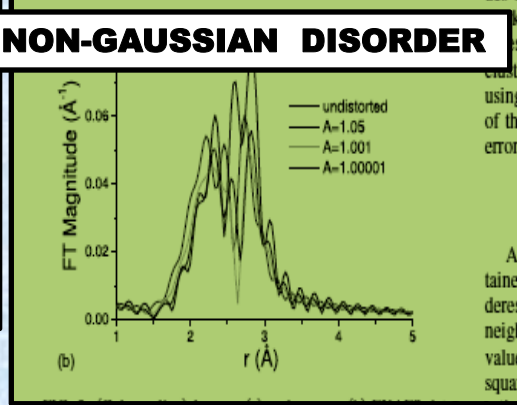
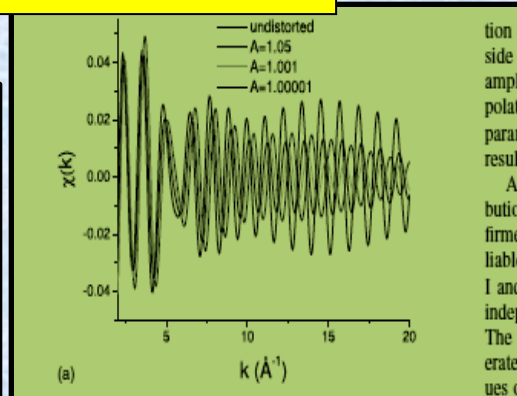
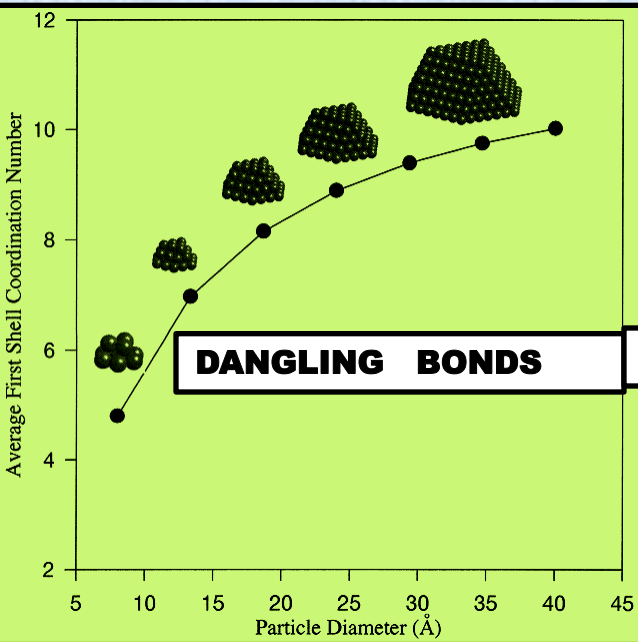
XAFS



EFFECTIVE

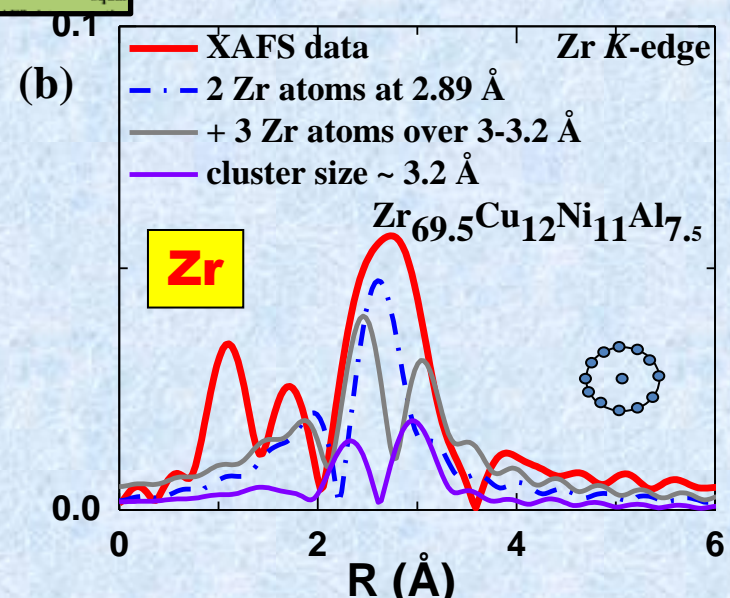
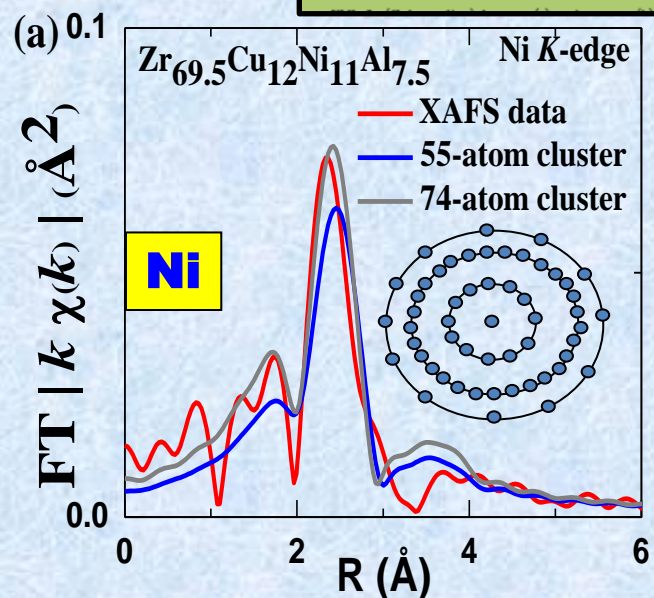
**COORDINATION NUMBER
(SITE-AVERAGE)**

CLUSTER SIZE / DISORDER

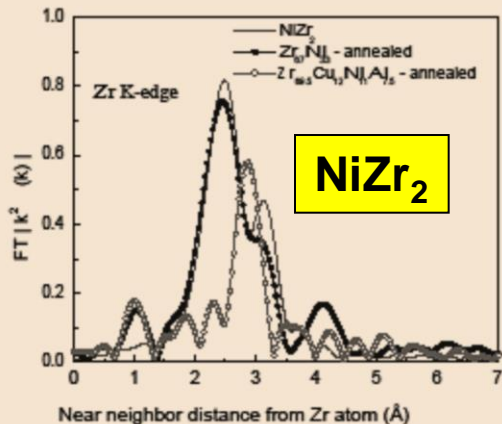


J. Phys. Chem. B
105 (51), 12699 (2001);
PRB 81, 115451 (2010)
... A.I FRENKEL

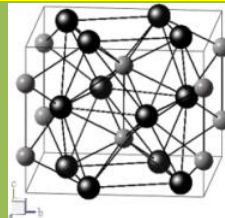
XAFS
+
AIMD



BINARY



> 2ND NEAREST NEIGHBOR



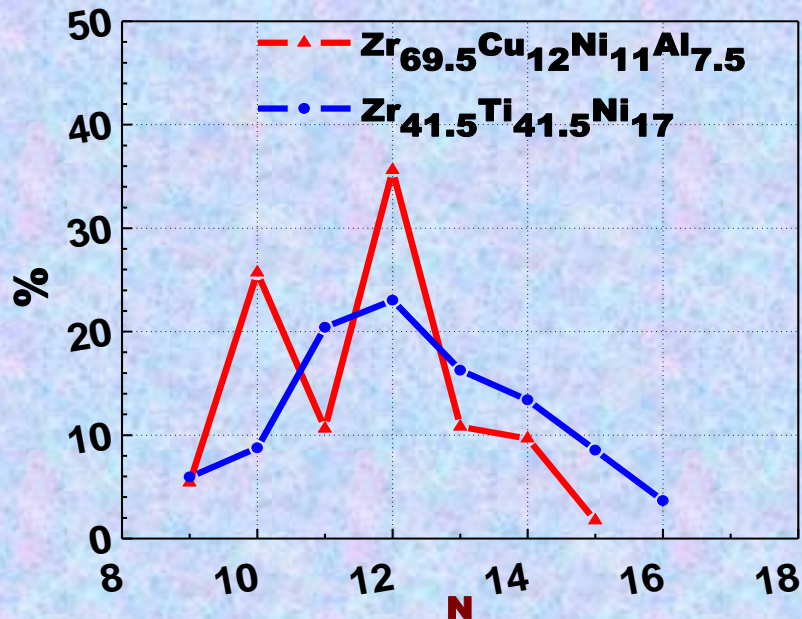
| | | |
|--|--------------------------------|--------------------------------|
| $R_{Zr-Ni} = 2.66 \text{ \AA}$ | $R_{Zr-Zr} = 3.05 \text{ \AA}$ | $R_{Zr-Zr} = 3.32 \text{ \AA}$ |
| NiZr₂: 2.76 \AA | 2.98-3.07 \AA | 3.36-3.43 \AA |

MULTI- component vs. BINARY

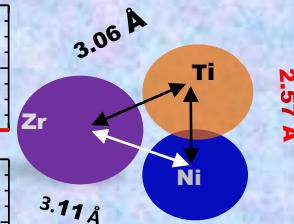
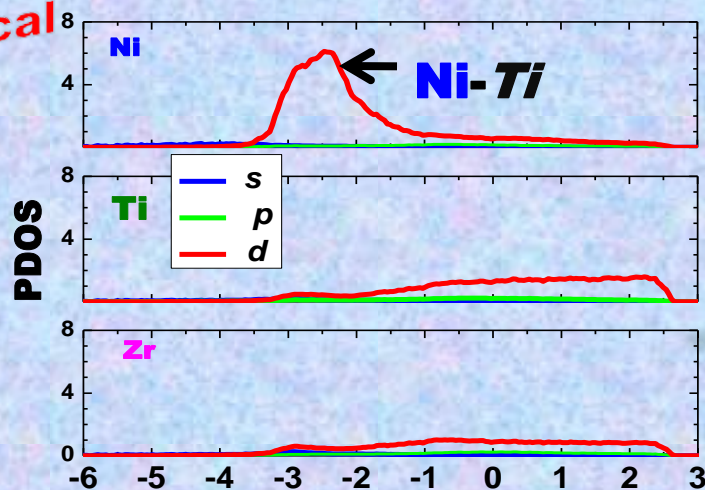
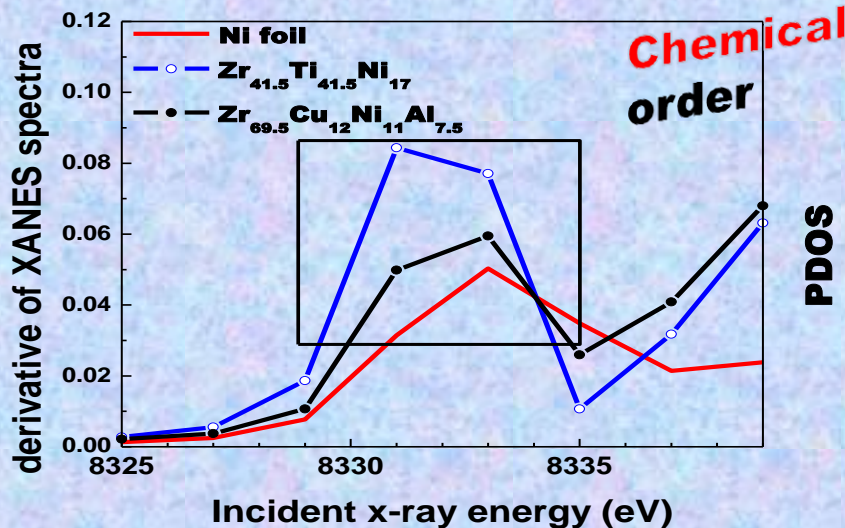
GOOD

POOR

| Cluster parameters | Correlation with GFA |
|--|-----------------------------|
| f_{ISRO} | + |
| Δf | + |
| σ^2 | - |
| $\Delta N_{\blacktriangleright}$ annealing | - |



| | Zr _{41.5} Ti _{41.5} Ni ₁₇ | Zr _{69.5} Cu ₁₂ Ni ₁₁ Al _{7.5} |
|----------------------|--|--|
| f_{ico} | 24% | 35% |
| Icosahedral fraction | | |
| Δf | 36% | 13% |
| Configuration | | |
| Diversity | | |



SUMMARY

MULTI-COMPONENT

BINARY



CHEMICAL ORDER

CONFIGURATION

ISRO >25%

XTAL

HETEROGENEITY

> 10%

0%

DEGREE OF ORDER

1st SHELL

> 2nd SHELL



UNDER INVESTIGATION

- **MEDIUM-RANGE-ORDER**
- **STRUCTURAL CHANGES**
 - *UNDER DEFORMATION*
 - *COOLING RATES*

ACKNOWLEDGEMENTS

- **HIGH PRESSURE & SYNCHROTRON RADIATION PHYSICS DIVISION - BARC**
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- **BESSY - GERMANY**
- **APS - USA**